

Microscopic origin for the orientation dependence of NV centers in chemical-vapor-deposited diamond

This content has been downloaded from IOPscience. Please scroll down to see the full text.

2014 J. Phys.: Condens. Matter 26 485004

(<http://iopscience.iop.org/0953-8984/26/48/485004>)

View [the table of contents for this issue](#), or go to the [journal homepage](#) for more

Download details:

IP Address: 162.105.243.68

This content was downloaded on 06/09/2016 at 15:44

Please note that [terms and conditions apply](#).

You may also be interested in:

[Stability and electronic structure of the low- grain boundaries in CdTe: a density functional study](#)

Ji-Sang Park, Joongoo Kang, Ji-Hui Yang et al.

[Impact of intrinsic atomic defects on the electronic structure of MoS2 monolayers](#)

Santosh KC, Roberto C Longo, Rafik Addou et al.

[Dimerization of boron dopant in diamond \(100\) epitaxy induced by strong pair correlation on the surface](#)

Xiaolong Yao, Yexin Feng, Zhenpeng Hu et al.

[Is there any substrate that is better than Ir\(100\) for diamond nucleation?](#)

Ling Liu and Lixin Zhang

[Substrate engineering in stabilizing epitaxial MgO\(111\) polar ultrathin films: first-principles calculations](#)

Feifei Li, Yexin Feng, Zhenpeng Hu et al.

[Interplay of hydrogen treatment and nitrogen doping in ZnO nanoparticles: a first-principles study](#)

Johann Gutjahr, Sung Sakong and Peter Kratzer

Microscopic origin for the orientation dependence of NV centers in chemical-vapor-deposited diamond

Yexin Feng^{1,2}, X Z Li², E G Wang², S B Zhang³ and Lixin Zhang¹

¹ School of Physics, Nankai University, Tianjin 300071, People's Republic of China

² School of Physics, Peking University, Beijing 100871, People's Republic of China

³ Department of Physics, Applied Physics, and Astronomy, Rensselaer Polytechnic Institute, Troy, NY 12180, USA

E-mail: lxzhang@nankai.edu.cn

Received 7 July 2014, revised 24 September 2014

Accepted for publication 6 October 2014

Published 24 October 2014

Abstract

First-principles calculation reveals that hydrogen, which is abundant in chemical vapor deposition (CVD), can significantly improve the uniformity of nitrogen-vacancy (NV) centers in diamond. It shows that the formation of NV centers can be described as a multi-step process: first, a substitutional N (N_C) is preferentially formed at the surface layer over that of either a carbon vacancy (V_C) or an in-plane nitrogen-vacancy-hydrogen (NVH) complex. Second, with the help of H, a V_C is preferentially incorporated in the newly formed topmost layer as a nearest neighbor to the N_C (now buried in the first sublayer). This NVH complex is even more stable than N_C on the same layer. Third, H protects the already formed NV centers by forming low-energy NVH_x complexes. These NV centers with their axes pointing along the directions of surface C–H bonds during their incorporation explain the experimental observations by CVD growth on (1 0 0) and (1 1 0) surfaces. Based on the model, we predict that CVD growth on (1 1 1) surface could eliminate the orientation domains to significantly improve the performance of NV centers.

Keywords: NV center, growth mechanism, surface, orientation dependence

(Some figures may appear in colour only in the online journal)

1. Introduction

Because of the unique spin and optical properties, nitrogen-vacancy (NV) center in diamond is a promising candidate for many attractive applications such as quantum information processing, magnetometry, and electric sensing [1–4]. Due to its quantum mechanical nature, manipulating the location, population, and orientation of these NV centers can be crucial for the success of such applications [5, 6]. Thus, it is paramount important to achieve a precise control of the formation of the NV centers in diamond. To date, nitrogen ion implantation and chemical vapor deposition (CVD) are the two widely deployed experimental methods for NV centers. The CVD method is believed to be superior to the nitrogen ion implantation method, because the latter will also cause uncontrollable crystal damages that seriously degrade the performance of the NV centers [7–10].

At present, the formation mechanism of NV centers by ion implantation has been generally agreed upon. First, the implanted N atoms incorporate onto the substitutional sites (N_C). With subsequent annealing ($>600^\circ$), the N_C 's capture nearby vacancies (V_C), which are highly mobile, to form the NV centers [8–13]. On the other hand, the incorporation of the NV centers in the CVD method is not yet well understood. Two possible mechanisms have been proposed [12, 13]: one is characterized by the capture of vacancy, similar to that of ion implantation, while the other focuses on the effect of surfaces where the NV centers form as a natural result of epitaxial growth. A number of experiments support the notion that the formation of the NV centers is a surface-related phenomenon [12, 14, 15]. In particular, it was reported that NV centers grown on diamond (1 1 0) surface show preferential orientations along only two of the four possible crystallographic axes [12, 14]. Interestingly, a

similar orientation preference was also observed for Si-related impurities grown in diamond [15]. In other words, only the interaction between diamond surfaces and impurities may explain the incorporation mechanism of NV centers in the CVD method. Besides, during the incorporation of the NV centers, other kinds of defects and impurities, such as N_C , V_C and nitrogen-vacancy-hydrogen (NVH) complexes, are also introduced [12–14, 16]. Unintentional defects/impurities could degrade the performance of the NV centers. Thus, understanding the fundamental physics of the impurities and maximizing the yield of NV centers during growth are crucially important for the applications of the NV center-based technologies.

In this paper, we propose a growth model for NV centers incorporated during CVD growth of diamond, based on first-principles calculations. A key factor of the model to nucleate and stabilize the NV centers is the presence of H. The model contains three steps: at first, an N_C is formed favorably on the top surface layer, substituting a surface CH group. In the subsequent growth, a V_C is preferably formed at the newly formed surface layer as the nearest neighbor to the N_C (now at the first sublayer), forming an NV center along one of the dangling bond directions of the surface. Thus, on the (100) surface, there are four possible orientations for the NV centers, whereas on the (110) and (111) surfaces, the respective symmetries determine that there are only two and one possible orientations for the NV centers, respectively. These surface-born NV centers are subsequently buried into diamond lattice with further deposition. The model applies to NV centers grown in diamond on all surfaces. Not only the results agree with available experimental observations on the (110) and (100) surfaces, it further predicts that, on the (111) surface, there should be only NV centers along one orientation, which could greatly improve the sensitivity of the grown NV centers.

2. Model and method

Spin-polarized density-functional theory calculations are performed with the Vienna Ab-initio simulation package (VASP) [17]. The Kohn–Sham wave functions are expanded in a plane wave basis set with a cut-off energy of 400 eV. The projector-augmented wave (PAW) method and PBE potential for the exchange-correlation functional are used [18]. A diamond surface is simulated by a slab with 4×4 surface cell, 12C layers, and a vacuum layer ≥ 12 Å. The bottom of the slab is passivated by H. The $2 \times 2 \times 1$ M-P k -point mesh is used, which has been shown to yield good convergence [19, 21]. All the atoms in the slab except in the two bottom layers are allowed to fully relax until the forces acting on them are less than 0.02 eV Å⁻¹.

An H-rich environment is typically adopted in the CVD growth [12, 20]. Thus, we consider H saturated surfaces in our calculations. The formation energy of an impurity or defect at the diamond surface is given by:

$$E_f = E_{\text{impurity}} - E_{\text{diamond}} + n\mu_N + m\mu_C + l\mu_H, \quad (1)$$

where E_{impurity} is the total energy of the slab with an impurity (or defect), E_{diamond} is the total energy of the reference slab without the impurity, and $n(m, l)$ is the number of N(C, H) atoms removed from the reference slab in order to incorporate the impurity. $\mu_N(\mu_C, \mu_H)$ is the atomic chemical potential of N(C, H). Under thermal equilibrium, μ_C should be the energy of C in bulk diamond. Since we consider the situation with maximum N concentration, μ_N can be taken as the energy of N in gas N₂. Depending on the growth environment, μ_H may vary but should be within a specific range, e.g. from -0.37 eV determined by the formation of CH₄ from diamond to 0 eV determined by the formation of H₂.

3. Results and discussion

3.1. Growth model on (110) surface

The (110) surface is a popular substrate for growing NV centers by CVD [12–14]. Under H-rich growth condition, each C on the surface is passivated by one H and the degree of relaxations of surface C are highly reduced after passivation. The fact that the surface C atoms are close in their positions to their bulk counterparts indicates that the surface is fully stabilized by H passivation [21].

During CVD growth, impurities are formed more easily at surface than in bulk, with qualitatively different formation mechanisms [22, 23]. We consider three types of impurities, N_C , V_C and NVH complex, as the most important near-surface precursors for NV incorporation. On the H saturated surface layer, the most stable N_C removes a C and a passivating H. The most stable V_C has all three C dangling bonds passivated by H. This is different from bulk where the limited space prevents the H passivation of all four dangling bonds. A surface NVH, which is the most likely precursor to the NV centers, on the other hand, is composed of an N_C , a V_C and an extra H. The atomic structures and formation energies of surface impurities are shown in figure 1.

In the bulk, the NV centers are usually charged depending upon the position of the Fermi level of the system. In the initial growth at the surfaces, the most favorable structures of the N-related defects are usually neutralized by hydrogen atoms. Thus, charged NV complexes are not considered in this study.

The formation energies of surface impurities are indeed noticeably lower than those in bulk ($E_f^{N_C} = 4.20$ eV, $E_f^{V_C} = 5.81$ eV and $E_f^{NV} = 5.42$ eV when $\mu_H = 0$ eV). This is an indication that the incorporation of impurities during CVD growth should be more effective. Among the three impurities/defects, V_C has the highest energy, N_C has the lowest, while NVH is in the middle. If we assume that the growth mode of diamond is layer-by-layer, we can expect that at the first step of the NV center incorporation, an N_C is formed favorably. Neither NVH nor V_C will form in the first step. In other words, the growth of NV center depends on further deposition.

Figure 2 shows the possible defect structures when the next layer is grown on the already N_C incorporated (110) surface. These structures either have low formation energies

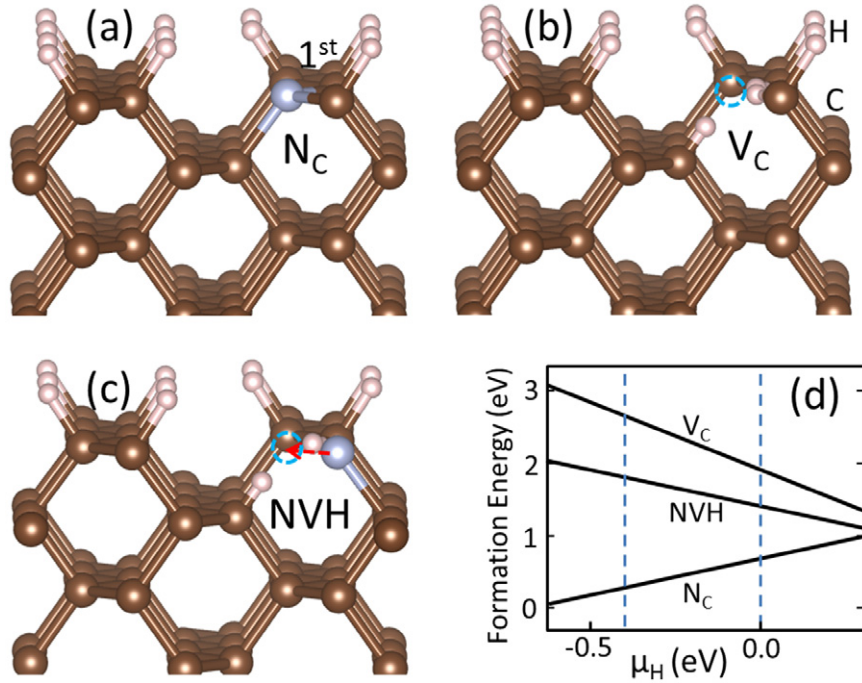


Figure 1. Atomic structures of (a) N_C (light blue ball is N), (b) V_C (dashed circle) and (c) an in-plane NVH complex at the top (110) surface layer, which is also the first layer of an epitaxial growth. Brown ball is carbon and smaller pink ball is hydrogen. Dashed arrow from N_C to V_C in the NVH complex denotes its orientation. (d) Impurity/Defect formation energy as a function of μ_H . Zero μ_H is for H in the state of H_2 , and -0.37 eV is for H in the state of CH_4 with μ_C equals that of bulk diamond.

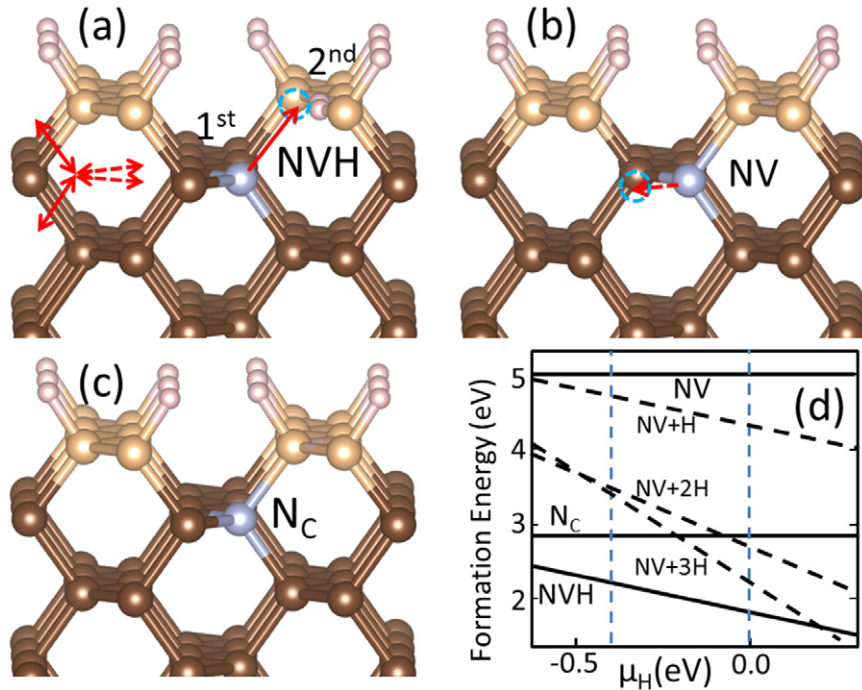


Figure 2. Atomic structures of the impurities when a second carbon layer (light brown balls) is deposited. (a) An out-of-plane NVH complex, (b) a buried in-plane NV center and (c) a buried N_C , both in the first layer beneath the second deposited layer. Solid and dashed arrows (both single and double headed) stand for out-of-plane dangling bond directions and in-plane bond directions of sp^3 diamond, respectively. (d) Defect formation energy as a function of μ_H . The dashed lines in (d) shows the formation energy change of the in-plane NV center with more passivating H atoms.

or are closely related to the formation of NV centers at the end. As shown in figure 2(a), the most stable structure is an NVH complex, composed of the N_C (now in the first sublayer) and a V_C in the newly grown surface layer. Typically, a surface layer containing no impurities (such as those in figures 2(b)

and (c)) is not energetically favored. Hence, horizontal NV center in the first sublayer (namely, figure 2(b)) is also not favored. Increasing the number of passivating H could reduce the energy for the complex in figure 2(b) but it is still much higher than the one in figure 2(a), as shown in figure 2(d).

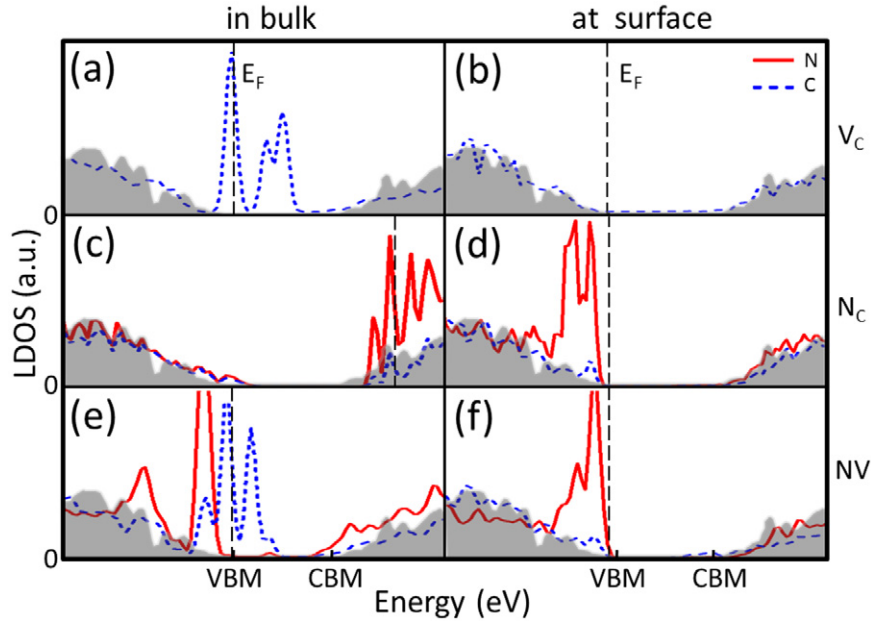


Figure 3. Local density of states (LDOS) of ((a) and (b)) V_C , ((c) and (d)) N_C and ((e) and (f)) NV center in bulk, and at (1 1 0) surface, respectively. The corresponding Fermi levels are indicated by vertical dashed lines. Shaded areas are the LDOS of bulk diamond, with the VBM and CBM indicated.

This result suggests that the most stable NVH complex at this stage must be formed in two steps: first, in depositing the first layer, an N substitutes a surface C. Second, with the deposition of the second layer, a V_C is formed in the newly formed surface layer as the nearest neighbor of the preceding N_C .

In figure 2(a), double arrows represent the four C–C bonds in diamond. On the (1 1 0) surface, two of them (the solid ones) give dangling bond directions out of the surface and the other two (dashed ones) are within the surface. Therefore, the discussion above asserts that possible orientations of the NVH complexes on the (1 1 0) surface are those along the solid arrows only. This result explains the observed preferential orientations of NV centers in [12]. It also indicates that near surface NVH complexes are responsible for the NV centers formed inside bulk (although the latter maybe only a fraction of the former).

To better understand the role of surface, figure 3 shows the local density of states (LDOS) of C and N in the vicinity of impurities in the bulk and at surface. We note that in the bulk, V_C has deep gap states originated from C dangling bonds; N_C is a relatively shallow donor with occupied anti-bonding states close to the conduction band minimum (CBM). These high-lying states are the reasons why isolated V_C and N_C have relatively high energies. Figure 3(e) shows that an NV center also consists of deep gap states but the N-related states are now near the valence band maximum (VBM). The net result of NV formation is thus to significantly lower the energy of occupied N states to lower system total energy. It explains why N_C attracts V_C in bulk diamond.

The situation is completely different at the surface. For the V_C , figure 3(b) shows that the deep gap states have completely gone. This is because the dangling bonds of the V_C have all been passivated by H. For N_C , figure 3(d) shows that the originally near-CBM states are now below the VBM. This is because the N_C has a dangling bond to accommodate the one

extra electron. The resulting surface NVH also has no gap states, as revealed by figure 3(f). It is this surface effect that makes the formation of N_C and V_C , as well as that of NVH complex in CVD diamond considerably more favorable. Note that figure 1(d) shows that a fully passivated V_C is still relatively high in energy than an N_C . This could be a strain effect as V_C has a limited space to host three H atoms.

Until now, the incorporation of NV centers is not completed yet. With more layers deposited, the atomic structure of the N-related impurities may still change, e.g. an incoming C could occupy a V_C site in NVH, thereby reducing the complex back to the original N_C . Figure 4 shows the representative atomic structures of the formed possible complexes, and the corresponding formation energies. We see that an NV center straddling the first and second sublayers without any H, has the highest energy. With passivating H (denoted as NVH_X , where $X = 1-3$ is the number of H), the energy could be lowered. Generally, the higher the H chemical potential, the lower the formation energies of NVH_X , among which NVH_2 is the most stable. When the H chemical potential is low, on the other hand, reduced N_C can have the lowest energy. These suggest that, in CVD growth, there is a delicate balance for the choice of μ_H to avoid having either too many hydrogenated NV centers or too many reduced N_C . Note that, unlike at the top surface, here the dangling bonds of NVH_2 are not fully passivated due to limited vacancy space. Experiments showed that NV centers in diamond are always accompanied by other defects, of which the most popular ones are N_C and NVH_X [12–14, 16] with a typical concentration ratio $N_C : NVH_X : NV = 300 : 30 : 1$. It qualitatively agrees with our calculated results at low μ_H in figure 4.

The properties of the NVH_X complexes have been studied experimentally [12, 13, 16]. Usually, hydrogenated NV centers are expected to contain only one H (bonded to a C). Commonly used experimental techniques for studying NV

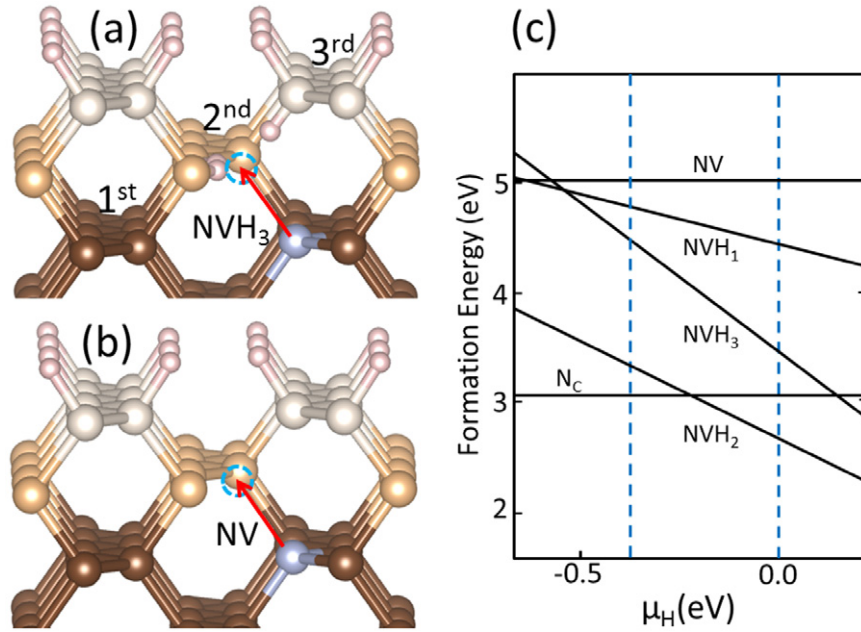


Figure 4. Atomic structures of the impurities when a third carbon layer (light gray balls) is deposited. (a) Out-of-plane NVH_3 complex and (b) out-of-plane NV center formed following the growth of the NVH complex in figure 2(a). (c) Defect formation energy as a function of μ_H .

centers usually involve electron paramagnetic resonance or optical spectroscopic techniques. However, these techniques are not so sensitive to local structures of the NVH_X . Based on the calculated formation energy, over a wide range of general growth conditions, we find $X = 2$ is most stable, followed by $X = 3$ and then by $X = 1$. The energy differences (~ 1 eV) are not small. Of course, at elevated temperature for diamond growth, one may also need to take into account the effect of entropy which would favor $X = 1$.

3.2. On (100) and (111) surfaces

Besides the (110), one may also use (100) or (111) surfaces for CVD growth [12–14, 19, 20]. For (100) surface, the most stable structure contains 2×1 dimer rows, in which every surface C binds to one H [23]. For (111) surface, the most stable structure is 1×1 where every surface C binds to one H [19]. Similarity of these hydrogenated surfaces suggests that the above (110) surface-assisted multi-step growth model for NV centers may also apply to (100) and (111) surfaces. Indeed, figures 5(a) and (c) show that an N_C can easily form at the topmost (100) and (111) surfaces similar to (110) surface. Figure 5(b) shows that, as the growth proceeds on the (100) surface, a vacancy can be easily formed adjacent to the N_C now at the first sublayer. The (111) surface is, however, slightly different, as growth may proceed in a double-layer-by-double-layer fashion. Figure 5(d) shows that under the double-layer growth mode, once the N_C is formed (now at the top part of the first sub bilayer), both NVH_X and NV could form in subsequent growth.

On the (100) surface, the dangling bonds have four directions, as shown by double arrows in figure 5(a). At any time, two equivalent directions appear on the top surface, but as the growth proceeds, the other two directions take over. This process repeats. Hence, over time, all four directions are

equivalent, so one can expect that there are four equivalent NV axes. This result agrees with experiment [14]. On the other hand, on the (111) surface, due to the doubly-layer growth mode, the direction of the dangling bonds is only along one axis, namely, along [111] at all times.

Thus, to incorporate the NV centers during CVD growth, on the (110) and (100) surfaces, one needs three separate steps, involving three sequential atomic layers. On the (111) surface, one needs, on the other hand, only two steps, involving two sequential atomic double-layers. Although detailed growth kinetics could be different on different surfaces to result in different concentration ratios among N_C , NV and NVH, the orientation dependence of the NV centers, which is analyzed here solely based on local energetic consideration with demonstrated success when compared with available experiments, should remain the same, irrespective of the kinetics.

NV centers with only one orientation have many advantages in terms of technological applications, such as the improvement of magnetic sensitivity [14]. Therefore, CVD growth of NV centers on (111) surface is highly desirable [24, 25]. The growth model developed here should also apply to other semiconductors, as a means to control the incorporation of desirable impurities. In this regard, we note that Si-related complexes in diamond also show surface-dependent properties [15].

4. Conclusion

In summary, first-principles study reveals the orientation dependent of NV centers in CVD grown diamond. In the multi-step growth model, the incorporation of the NV centers is enhanced by H. Despite the simplicity of the model, it explains the formation of NV centers with four possible orientations in (100)-grown diamond and two possible

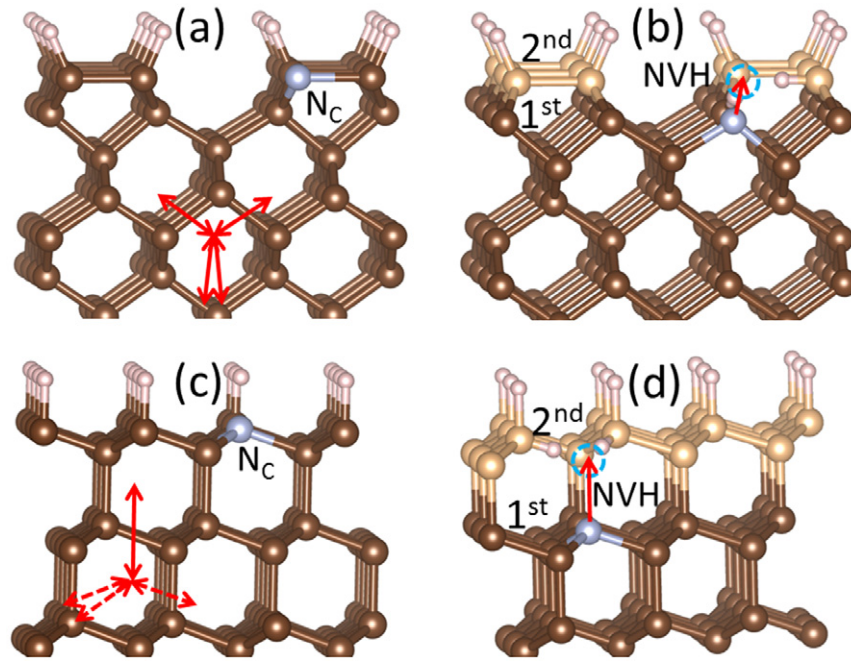


Figure 5. Atomic structures of (a) an N_C and (b) an NVH complex on the (1 0 0) surface and (c) an N_C and (d) an NVH complex on the (1 1 1) surface. Solid arrows indicate the possible orientations of the NV centers, while dashed arrows indicate the impossible ones. The growth sequence is denoted by first and second single (for (0 0 1)) and double (for (1 1 1)) layers.

orientations in (1 1 0)-grown diamond. These results agree with available experiments, demonstrating the validity of the model. Given the similarities in the NV center formation and incorporation near surfaces, we propose that using the (1 1 1) surface, one should be able to grow NV centers with only one orientation, which could be a significant step forward toward fundamental study of the NV centers as well as for technological applications.

Acknowledgments

The work is supported by the NSF of China with grant nos. 11074128, 11274179 and 91227101, and National 973 projects of China with nos. 2011CB922102 and 2012CB921900. SBZ was supported by the Department of Energy under grant no. DE-SC0002623.

References

- [1] Grinolds M S, Maletinsky P, Hong S, Lukin M D, Walsworth R L and Yacoby A 2011 *Nature Phys.* **7** 687
- [2] Dolde F *et al* 2011 *Nature Phys.* **7** 459
- [3] Bar-Gill N, Pham L M, Belthangady C, Le Sage D, Cappellaro P, Maze J R, Lukin M D, Yacoby A and Walsworth R L 2012 *Nature Commun.* **3** 858
- [4] Mamin H J, Kim M, Sherwood M H, Rettner C T, Ohno K, Awschalom D D and Rugar D 2013 *Science* **339** 557
- [5] Stacey A *et al* 2012 *Adv. Mater.* **24** 3333
- [6] Barclay P E, Fu K M C, Santori C and Beausoleil R G 2009 *Appl. Phys. Lett.* **95** 191115
- [7] Ohno K, Heremans F J, Bassett L C, Myers B A, Toyli D M, Bleszynski-Jayich A C, Palmstrøm C J and Awschalom D D 2012 *Appl. Phys. Lett.* **101** 082413
- [8] Santori C, Barclay P E, Fu K M C and Beausoleil R G 2009 *Phys. Rev. B* **79** 125313
- [9] Staudacher T, Ziem F, Häussler L, Stöhr R, Steinert S, Reinhard F, Scharpf J, Denisenko A and Wrachtrup J 2012 *Appl. Phys. Lett.* **101** 212401
- [10] Pezzagna S, Naydenov B, Jelezko F, Wrachtrup J and Meijer J 2010 *New J. Phys.* **12** 065017
- [11] Botsoa J, Sauvage T, Adam M P, Desgardin P, Leoni E, Courtois B, Treussart F, and Barthe M F 2011 *Phys. Rev. B* **84** 125209
- [12] Edmonds A M, D'Haenens-Johansson U F S, Cruddace R J, Newton M E, Fu K M C, Santori C, Beausoleil R G, Twitchen D J and Markham M L 2012 *Phys. Rev. B* **86** 035201
- [13] Glover C, Newton M E, Martineau P, Twitchen D J and Baker J M 2003 *Phys. Rev. Lett.* **90** 185507
- [14] Pham L M, Bar-Gill N, Le Sage D, Belthangady C, Stacey A, Markham M, Twitchen D J, Lukin M D and Walsworth R L 2012 *Phys. Rev. B* **86** 121202
- [15] D'Haenens-Johansson U F S, Edmonds A M, Newton M E, Goss J P, Briddon P R, Baker J M, Martineau P M, Khan R U A, Twitchen D J and Williams S D 2010 *Phys. Rev. B* **82** 155205
- [16] Goss J P, Briddon P R, Jones R and Sque S 2003 *J. Phys.: Condens. Matter* **15** S2903
- [17] Kresse G and Hafner J 1993 *Phys. Rev. B* **47** R558
- [18] Perdew J P, Burke K and Ernzerhof M 1996 *Phys. Rev. Lett.* **77** 3865
- [19] Kern G, Hafner J and Kresse G 1996 *Surf. Sci.* **366** 445
- [20] Chu C J, Hauge R H, Margrave J L and D'Evelyn M P 1992 *Appl. Phys. Lett.* **61** 1393
- [21] Kern G and Hafner J 1997 *Phys. Rev. B* **56** 4203
- [22] Yao X, Feng Y, Hu Z, Zhang L and Wang E G 2013 *J. Phys.: Condens. Matter* **25** 045011
- [23] Winn M D, Rassinger M and Hafner J 1997 *Phys. Rev. B* **55** 5364
- [24] Doherty M W, Michl J, Dolde F, Jakobi I, Neumann P, Manson N B and Wrachtrup J 2014 *New J. Phys.* **16** 063067
- [25] Fukui T, Doi Y, Miyazaki T, Miyamoto Y, Kato H, Matsumoto T, Makino T, Yamasaki S, Morimoto R and Tokuda N 2014 *Appl. Phys. Express* **7** 055201